## 國立彰化師範大學 98 學年度碩士班招生考試試題

糸所:電子工程學系

組別: 甲 組

科目:(乙)近代物理

☆☆請在答案紙上作答☆☆

Physical constants:

共2頁,第1頁

Planck's constant h =  $6.626 \times 10^{-34}$  J-s., c =  $3 \times 10^{8}$  m/sec, m<sub>e</sub> =  $9.1 \times 10^{-31}$  kg. e =  $1.602 \times 10^{-19}$  C,  $\varepsilon_{0} = 8.854 \times 10^{-12}$  C<sup>2</sup>/N·m<sup>2</sup>, The Boltzmann's constant k<sub>B</sub> =  $1.38 \times 10^{-23}$  J/K.

- Electron beams are used for double slit interference experiment. A (B) is the intensity profile when only top (bottom) slit is open. Explain why the resultant intensity profile is NOT A+B. (see Fig.1) (10%)
- Qualitatively draw the wave functions (n = 4 & n = 5) of the symmetric infinite potential well shown in Fig. 2. (10%)
- 3. Solve the time independent Schrodinger equation for an electron in an infinite potential well with width L. What are the wavefunctions and energies of the lowest two modes? (20%)
- 4. (a) In Bohr's H-atom model, what is quantized angular momentum assumption? (5%)
  - (b) Calculate the quantized energy levels in H-atom. (10%)
  - (c) Verify that when the electron transits from higher energy level to lower energy level, the

frequency of the photon it emits is  $f = \frac{E_1}{h} (\frac{1}{n_f^2} - \frac{1}{n_i^2}) . (5\%)$ 

- (d) What is the frequency range of Balmer series spectrum lines of H-atom? (5%)
- 5. Explain why the Auger electron spectrum for elemental Al and Al oxide are shifted. (see Fig. 3) (10%)
- (a) 0.012 nm X-ray serves as the photon in the Compton effect experiment. Find the maximum wavelength of the scattered photon. (10%)
  - (b) Find the maximum kinetic energy of the scattered electron could have. (5%)
- The potential energies are drawn down to scale. Compare the ground state energy (n=1) in each case. List them from highest to lowest. (see Fig. 4) (10%)

*The figures are in page 2!* 

## 國立彰化師範大學 98 學年度碩士班招生考試試題

系所:電子工程學系

組別: 甲 組

科目:(乙)近代物理

☆☆請在答案紙上作答☆☆

共2頁,第2頁

